



PATENT
Attorney Docket No. 223982
Client Reference No. GN 02060 OCONPEDOTDIOXIDE

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Application of:

Groenendaal et al.

Application No. 10/649,761

Filed: August 27, 2003

Art Unit: Unassigned

Examiner: Unassigned

For: NEW 3,4-ALKYLENEDIOXYTHIOPHENEDIOXIDE
COMPOUNDS AND POLYMERS COMPRISING
MONOMERIC UNITS THEREOF

CLAIM OF PRIORITY

Commissioner for Patents
P.O. Box 1450
Alexandria, VA 22313-1450

Dear Sir:

In accordance with the provisions of 35 USC 119, Applicants claim the priority of the following application:

Application No. PCT/EP02/09887, filed in the European Patent
Office on September 2, 2002.

A certified copy of the above-listed priority document is enclosed.

Respectfully submitted,

Christopher T. Griffith, Reg. No. 33,392
LEYDIG, VOIT & MAYER, LTD.
Two Prudential Plaza, Suite 4900
180 North Stetson
Chicago, Illinois 60601-6780
Telephone: (312) 616-5600
Facsimile: (312) 616-5700

Date: October 1, 2003

CERTIFICATE OF MAILING

I hereby certify that this CLAIM OF PRIORITY (along with any documents referred to as being attached or enclosed) is being deposited with the United States Postal Service on the date shown below with sufficient postage as first class mail in an envelope addressed to: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

Date: 10/1/03



Bescheinigung

Certificate

Attestation

Die angehefteten Unterlagen stimmen mit der ursprünglich eingereichten Fassung der auf dem nächsten Blatt bezeichneten internationalen Patentanmeldung überein.

The attached documents are exact copies of the international patent application described on the following page, as originally filed.

Les documents fixés à cette attestation sont conformes à la version initialement déposée de la demande de brevet international spécifiée à la page suivante.

Den Haag, den
The Hague,
La Haye, le

27. 08. 2003

Der Präsident des Europäischen Patentamts
Im Auftrag
For the President of the European Patent Office
Le Président de l'Office européen des brevets
p.o.


Mme. C. Rossi

Patentanmeldung Nr.
Patent application n° PCT/EP 02/09887
Demande de brevet n°

Blatt 2 d r Bescheinigung
Sheet 2 of the certificate
Page 2 de l'attestation



Anmeldung Nr.: PCT/EP 02/09887
Application no.:
Demande n°:

Anmelder:
Applicant(s):
Demandeur(s):

1. AGFA - GEVAERT - Mortsel, Belgium
2. GROENENDAAL, Bert - Mortsel, Belgium (US only)
3. LOUWET, Frank - Mortsel, Belgium (US only)

Bezeichnung der Erfindung: New 3, 4-alkylenedioxythiophenedioxide compounds and polymers comprising
Title of the invention: monomeric units thereof
Titre de l'invention:

Anmeldetag: 02 September 2002 (02.09.2002)
Date of filing:
Date de dépôt:

In Anspruch genommene Priorität(en)
Priority(ies) claimed
Priorité(s) revendiquée(s)

Staat:
State:
Pays:

Tag:
Date:
Date:

Aktenzeichen:
File no.
Numéro de dépôt:

Benennung von Vertragsstaaten : Siehe Formblatt PCT/RO/101 (beigefügt)
Designation of contracting states : See Form PCT/RO/101 (enclosed)
Désignation d'états contractants : Voir Formulaire PCT/RO/101 (ci-joint)

Bemerkungen: Further applicants:
R marks:
Remarques: 4. ANDRIESSEN, Hieronymus - Mortsel, Belgium (US only)

PCT REQUEST

GN02060

Original (for SUBMISSION) - printed on 29.08.2002 01:46:34 PM

V	D signati n f States	
V-1	Regional Patent (other kinds of protection or treatment, if any, are specified between parentheses after the designation(s) concerned)	<p>AP: GH GM KE LS MW MZ SD SL SZ TZ UG ZM ZW and any other State which is a Contracting State of the Harare Protocol and of the PCT</p> <p>EA: AM AZ BY KG KZ MD RU TJ TM and any other State which is a Contracting State of the Eurasian Patent Convention and of the PCT</p> <p>EP: AT BE BG CH&LI CY CZ DE DK EE ES FI FR GB GR IE IT LU MC NL PT SE SK TR and any other State which is a Contracting State of the European Patent Convention and of the PCT</p> <p>OA: BF BJ CF CG CI CM GA GN GQ GW ML MR NE SN TD TG and any other State which is a member State of OAPI and a Contracting State of the PCT</p>
V-2	National Patent (other kinds of protection or treatment, if any, are specified between parentheses after the designation(s) concerned)	<p>AE AG AL AM AT AU AZ BA BB BG BR BY BZ CA CH&LI CN CO CR CU CZ DE DK DM DZ EC EE ES FI GB GD GE GH GM HR HU ID IL IN IS JP KE KG KP KR KZ LC LK LR LS LT LU LV MA MD MG MK MN MW MX MZ NO NZ OM PH PL PT RO RU SD SE SG SI SK SL TJ TM TN TR TT TZ UA UG US UZ VC VN YU ZA ZM ZW</p>
V-5	Precautionary Designation Statement In addition to the designations made under items V-1, V-2 and V-3, the applicant also makes under Rule 4.9(b) all designations which would be permitted under the PCT except any designation(s) of the State(s) indicated under item V-6 below. The applicant declares that those additional designations are subject to confirmation and that any designation which is not confirmed before the expiration of 15 months from the priority date is to be regarded as withdrawn by the applicant at the expiration of that time limit.	
V-6	Exclusion(s) from precautionary designations	NONE
VI	Priority claim	NONE
VII-1	International Searching Authority Chosen	European Patent Office (EPO) (ISA/EP)

27/08/2002 10:05

1

NEW 3,4-ALKYLENEDIOXYTHIOPHENEDIOXIDE COMPOUNDS AND POLYMERS
COMPRISING MONOMERIC UNITS THEREOF

Field of the invention

5

The present invention concerns new 3,4-alkylenedioxythiophene-dioxide compounds and polymers comprising monomeric units thereof.

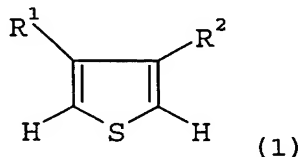
Background of the invention.

10

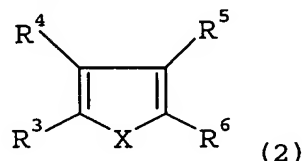
In 1970, Mock in Journal of the American Chemical Society, volume 92, page 7610 disclosed 2,5-di-tert-butyl-thiophene-1,1-dioxide. In 1992, Furukawa et al. in Heterocycles, volume 34, pages 1085-1088, disclosed 2,5-bis(trimethylsilyl)-thiophene-1,1-dioxide. In 1998, Barbarella et al. in Journal of Organic Chemistry, volume 63, page 5497, disclosed 2-(dimethyl-tert-butyl-silyl)-5-bromothiophene-1,1-dioxide, 2-n-hexyl-5-bromo-thiophene-1,1-dioxide, 2,5-dibromo-3,4-dihexyl-thiophene-1,1-dioxide and 2,5-di-n-hexyl-thiophene-1,1-dioxide. EP-A 1 041 132 discloses a luminescent organic material for light-emitting devices, characterized by comprising at least one thienyl-S,S-dioxide unit. Specific thiophene-1,1-dioxide compounds are disclosed: 2,5-diphenyl-thiophene-1,1-dioxide, 3,4-dihexyl-2,5-diphenyl-thiophene-1,1-dioxide, 3,4-dineopentyl-2,5-diphenyl-thiophene-1,1-dioxide, 3,4-dimethyl-2,5-diphenyl-thiophene-1,1-dioxide, 3,4-dihexyl-2,5-bis(diphenyl)-thiophene-1,1-dioxide and 2,5-trimethylsilyl-thiophene-1,1-dioxide.

Polythiophenes have been studied extensively due to their interesting electrical and/or optical properties. Polythiophenes become electrically conducting upon chemical or electrochemical oxidation or reduction.

EP-A 257 573 discloses an intrinsically electrically conductive polymer, wherein through connection in the 2-position and/or the 5-position are coupled to one another, statistically averaged from 60 to 100% by weight structural units, which are derived from at least one monomer of the formula (1):

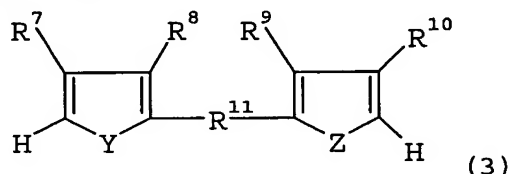


in which R^1 is a C1-C2-alkoxy group or $-O(CH_2CH_2O)_nCH_3$ with $n = 1$ to 4 and R^2 is a hydrogen atom, a C₁₋₁₂-alkyl group, a C₁₋₁₂-alkoxy group or $-O(CH_2CH_2O)_nCH_3$ with $n = 1$ to 4, or R^1 and R^2 together are $-O(CH_2)_m-CH_2-$ or $-O(CH_2)_m-O-$ with $m = 1$ to 12,
 5 0 to 40% by weight structural units, which are derived from at least one monomer of the formula (2):

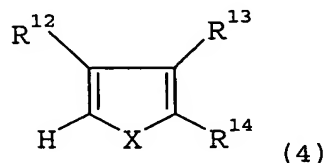


wherein R^4 and R^5 are independently of one another a hydrogen atom, a halogen atom, a C₁₋₁₂-alkyl group or aryl or together with C-atoms
 10 connected to them form an aromatic ring, R^3 and R^6 independently of one another represent a hydrogen atom or R^3 together with R^4 and the C-atoms connected to them or R^5 together with R^6 and the C-atoms connected to them each form an aromatic ring, X represents an oxygen atom, a sulphur atom, a =NH group, a =N-alkyl group or a =N-aryl
 15 group,

0 to 40% by weight structural units, which are derived from at least one monomer of formula (3):



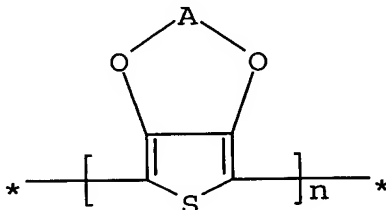
where R^7 , R^8 , R^9 and R^{10} independently of one another represent a
 20 hydrogen atom, a C₁₋₁₂-alkyl group, a C₁₋₁₂-alkoxy group or an aryl group, Y and Z independently of one another represent an oxygen atom, a sulphur atom, a =NH group, a =N-alkyl group or a =N-aryl group, R^{11} represents an arylene group, a heteroarylene group or a conjugated system of the formula $(CH=CH)_o$, wherein o is 1, 2 or 3, 0
 25 to 40% by weight structural units, which are derived from at least one monomer of formula (4):



wherein R^{12} and R^{13} independently of one another represent a hydrogen atom, a halogen atom, a C₁₋₁₂-alkyl group, a C₁₋₁₂-alkoxy group, a C₁₋₄-alkylamino group or a C₁₋₄-acylamino group, R^{14} represents a
 30 halogen atom, a C₁₋₁₂-alkyl group, a C₁₋₁₂-alkoxy group, a C₁₋₄-alkylamino group or a C₁₋₄-acylamino group and X has the meaning

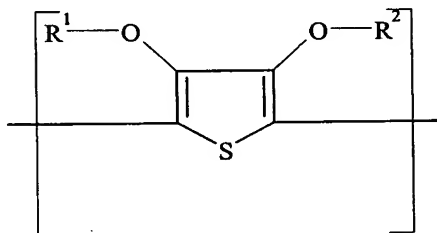
given above, wherein the polymer in the oxidized form is completely soluble in dipolar aprotic solvents at 25°C and solutions with a content of at least 0.1g of the polymer in 100 mL solvent at 25°C are obtained.

5 EP-A 339 340 discloses a polythiophene containing structural units of the formula:



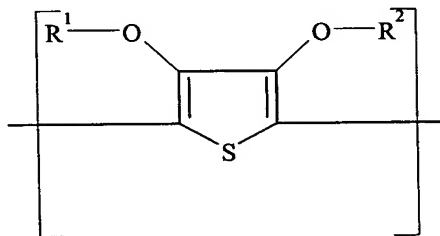
in which A denotes an optionally substituted C₁₋₄-alkylene group and its preparation by oxidative polymerization of the corresponding
10 thiophene.

EP-A 440 957 discloses dispersions of polythiophenes, constructed from structural units of formula (I):



in which R¹ and R² independently of one another represent hydrogen or
15 a C₁₋₄-alkyl group or together form an optionally substituted C₁₋₄-alkylene residue, in the presence of polyanions.

EP-A-686 662 discloses mixtures of A) neutral polythiophenes with the repeating structural unit of formula (I),



20 in which R¹ and R² independently of one another represent hydrogen or a C₁₋₄-alkyl group or together represent an optionally substituted C₁₋₄-alkylene residue, preferably an optionally with alkyl group substituted methylene, an optionally with C₁₋₁₂-alkyl or phenyl group substituted 1,2-ethylene residue or a 1,2-cyclohexene residue, and
25 B) a di- or polyhydroxy- and/or carboxy groups or amide or lactam group containing organic compound; and conductive coatings therefrom

which are tempered at elevated temperature, preferably between 100 and 250 °C, during preferably 1 to 90 seconds to increase their resistance preferably to < 300 ohm/square.

PEDOT/PSS layers have been widely used in light emitting diodes
5 as an intermediate layer between a indium tin dioxide layer and a hole transporting layer. However, the conductivity of such layers is too high and so-called cross-talk arises resulting in short-circuiting.

There is a need for hole-conducting intermediate layers with a
10 similar work function, but reduced electrical conductivity.

Objects of the invention.

It is therefore an aspect of the present invention to provide
15 polymers which exhibit high visible light transmittances, good processability and similar work functions but reduced electrical conductivities compared with poly(3,4-alkylenedioxythiophene)'s.

It is also a further aspect of the present invention to provide an electroconductive pattern.

20 Further aspects and advantages of the invention will become apparent from the description hereinafter.

Summary of the invention

25 It has been surprisingly found that copolymerizing EDOT-dioxide with EDOT reduces the molecular weight of copolymer produced in the presence of PSS. It surprisingly appears that EDOT-dioxide may act as a chain stopper in the copolymerization process resulting in a reduced molecular weight, reduced conjugation length and reduced
30 conductivity, although the work function as measured by UPS measurements is unaffected. This makes this copolymer a very interesting candidate to replace PEDOT/PSS-layers in organic light emitting diodes so-called OLED's due to so-called reduced cross-talk i.e. lateral electrical conductivity.

35 Aspects of the present invention are realized with a 3,4-alkylenedioxythiophenedioxide compound.

Aspects of the present invention are also provided by a polymer comprising monomeric units of at least one of the above-mentioned compounds.

40 Aspects of the present invention are also provided by an aqueous dispersion of a polymer comprising monomeric units of at least one of the above-mentioned compounds and a polyanion.

Aspects of the present invention are also provided by a chemical polymerization process for preparing the above-mentioned aqueous dispersion comprising the steps of: (i) providing a solution of a polyanion; (ii) adding the above-mentioned compound and a thiophene or a pyrrole compound to the solution provided in step (i); and (iii) adding an oxidizing or reducing system to the mixture provided in step (ii).

Aspects of the present invention are also provided by the use of the above-mentioned dispersion for coating an object.

Aspects of the present invention are also provided by a printable paste containing the above-mentioned aqueous dispersion.

Aspects of the present invention are also provided by an electroconductive layer containing the above-mentioned polymer.

Aspects of the present invention are also provided by the use of the above-mentioned electroconductive layer in a light emitting diode.

Aspects of the present invention are also provided by an antistatic layer containing the above-mentioned polymer.

Aspects of the present invention are also provided by an electroconductive pattern comprising a polymer containing 3,4-alkylenedioxythiophene monomer units and a polymer containing 3,4-alkylenedioxythiophenedioxide monomer units.

Aspects of the present invention are also provided by a process for preparing the above-mentioned electroconductive pattern comprising the steps of: providing a layer of a polymer containing 3,4-alkylenedioxythiophene monomer units; and pattern-wise oxidizing the polymer containing 3,4-alkylenedioxythiophene monomer units to polymers containing 3,4-alkylenedioxythiophenedioxide monomer units.

Further aspects of the present invention are disclosed in the dependent claims.

Detailed description of the invention.

Figure 1 is a schematic representation of a side view and a top view of the layer configuration used in devices 1 to 3 and the circuit used for obtaining electroluminescence in which:

- A represents a poly(ethylene terephthalate) support;
- B represents a sputtered ITO layer;
- C represents an electron blocking layer;
- D represents an electroluminescent layer containing ZnS:Cu nanoparticles and a binder;

- E represents an evaporated aluminium electrode
F represents conductive silver paste dots for contacting
G represents IV-power source (Power Supply ES 030-5 of Delta Elektronika)
H represents electric conductive copper wires
I indicates the first layer
II indicates the second layer
III indicates the third layer

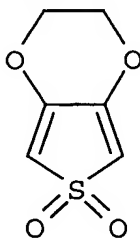
Definitions

The term C₁₋₅-alkylene group represents methylenedioxy, 1,2-
5 ethylenedioxy, 1,3-propylenedioxy, 1,4-butylenedioxy and 1,5-pentylenedioxy groups.

The term alkyl means all variants possible for each number of carbon atoms in the alkyl group i.e. for three carbon atoms: n-propyl and isopropyl; for four carbon atoms: n-butyl, isobutyl and
10 t-butyl; for five carbon atoms: n-pentyl, 1,1-dimethyl-propyl, 2,2-dimethylpropyl and 2-methyl-butyl etc.

The term 3,4-alkylenedioxythiophene compound as used in disclosing the present invention means a thiophene molecule substituted with an optionally substituted oxy-alkyl-oxy bridge
15 between the 3- and 4-positions thereof.

The term 3,4-alkylenedioxythiophenedioxide compound as used in disclosing the present invention means a thiophene molecule in which the sulphur atom is directly bonded to two oxygen atoms and in which the thiophene molecule is substituted with an optionally substituted
20 oxy-alkyl-oxy bridge between the 3- and 4-positions thereof e.g. 2,3-dihydro-thieno[3,4-b][1,4]dioxine 6,6-dioxide:



The term polymer as used in disclosing the present invention includes homopolymers, copolymers, terpolymers, graft copolymers and
25 block copolymers.

The term aqueous dispersion of a polymer means that particles of the polymer are suspended in the aqueous medium.

The term aqueous for the purposes of the present invention means containing at least 60% by volume of water, preferably at least 80% by volume of water, and optionally containing water-miscible organic solvents such as alcohols e.g. methanol, ethanol, 2-propanol, 5 butanol, iso-amyl alcohol, octanol, cetyl alcohol etc.; glycols e.g. ethylene glycol; glycerine; N-methyl pyrrolidinone; methoxypropanol; and ketones e.g. 2-propanone and 2-butanone etc.

The term conductive layer as used in disclosing the present invention includes both electroconductive coatings and antistatic 10 layers.

The term electroconductive means having a surface resistance below $10^6 \Omega/\text{square}$.

The term antistatic means having a surface resistance in the range from 10^6 to $10^{11} \Omega/\text{square}$ meaning it cannot be used as an 15 electrode.

The term "conductivity enhancement" refers to process disclosed in EP-A 686 662 in which the conductivity is enhanced e.g. by contact with high boiling point liquids such as di- or polyhydroxy- and/or carboxy groups or amide or lactam group containing organic 20 compound optionally followed by heating at elevated temperature, preferably between 100 and 250°C, during preferably 1 to 90 seconds, results in conductivity increase. Alternatively in the case of aprotic compounds with a dielectric constant ≥ 15 , e.g. N-methyl-pyrrolidinone, temperatures below 100°C can be used as disclosed in 25 EP-A 1 003 179. Such conductivity enhancement is observed with polythiophenes and can take place during the preparation of the outermost layer or subsequently. Particularly preferred liquids for such treatment are N-methyl-pyrrolidinone and diethylene glycol such as disclosed in EP-A 686 662 and EP-A 1 003 179.

30 PEDOT as used in the present disclosure represents poly(3,4-ethylenedioxythiophene).

EDOT as used in the present disclosure represents 3,4-ethylenedioxythiophene.

ADOT as used in the present disclosure represents 3,4-35 alkylendioxythiophene.

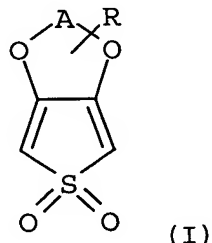
ADOT-dioxide as used in the present disclosure represents 3,4-alkylendioxythiophenedioxide.

PSS as used in the present disclosure represents poly(styrenesulphonic acid) or poly(styrenesulphonate).

40 PET as used in the present disclosure represents poly(ethylene terephthalate).

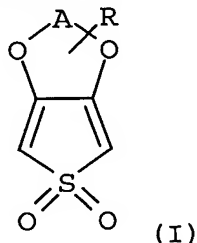
3,4-Alkylenedioxythiophenedioxide compound

According to a first embodiment of the 3,4-alkylenedioxy-thiophenedioxide compound, according to the present invention, the
 5 3,4-alkylenedioxythiophenedioxide compound is represented by formula (I):



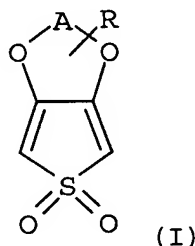
in which: A represents a C₁₋₅-alkylene bridge; R represents an optionally substituted C₁₋₂₄-alkyl, C₃₋₁₈-cycloalkyl, C₁₋₁₈-alkoxy or
 10 polyethylene oxide group (optionally with at least one substituent selected from the group consisting of an alcohol, amide, ether, ester or sulphonate group), an optionally substituted aryl group or an optionally substituted acyl group.

According to a second embodiment of the 3,4-alkylenedioxy-thiophenedioxide compound, according to the present invention, the
 15 3,4-alkylenedioxythiophene compound is represented by formula (I)



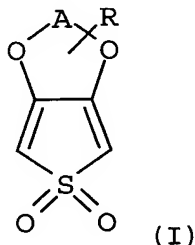
in which: A represents a C₁₋₅-alkylene bridge; and R represents an optionally substituted C₆₋₂₀-alkyl group.

According to a third embodiment of the 3,4-alkylenedioxy-thiophenedioxide compound, according to the present invention, the
 20 3,4-alkylenedioxythiophenedioxide compound is represented by formula (I):



25 in which: A represents a C₁₋₅-alkylene bridge; and R represents an optionally substituted C₈₋₁₈-alkyl group.

According to a fourth embodiment of the 3,4-alkylenedioxythiophenedioxide compound, according to the present invention, the 3,4-alkylenedioxythiophenedioxide compound is represented by formula (I):



5 in which: A represents a C₁₋₅-alkylene bridge; and R represents a substituted C₂₋₂₄-alkyl, C₃₋₁₈-cycloalkyl, C₁₋₁₈-alkoxy or aryl group containing at least one substituent selected from the group consisting of an ether, an ester, an amide, a sulphonate, a
10 phosphonate, a halogen and a hydroxy group.

According to a fifth embodiment of the 3,4-alkylenedioxythiophenedioxide compound, according to the present invention, the 3,4-alkylenedioxythiophenedioxide compound is 2,3-dihydrothieno[3,4-b][1,4]dioxine 6,6-dioxide.

15 3,4-Alkylenedioxythiophenedioxide compounds can be prepared from the corresponding thiophene compounds by oxidation by peroxides such as m-chloroperoxybenzoic acid (m-CPBA), dimethyldioxirane and HOF·CH₃CN [see references in Journal of Organic Chemistry, volume 63, pages 5497-5506 (1998)].

20 3,4-Alkylenedioxythiophenedioxide compounds were found not to homopolymerize either electrochemically or chemically.

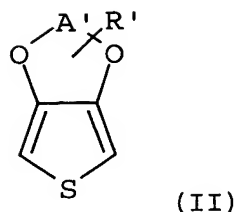
Polymer comprising 3,4-alkylenedioxythiophenedioxide monomer units

25 According to a first embodiment of the polymer, according to the present invention, the polymer further comprises monomeric units of at least one thiophene or pyrrole compound.

According to a second embodiment of the polymer, according to the present invention, the polymer further comprises monomeric units
30 of at least one 3,4-alkylenedioxythiophene compound.

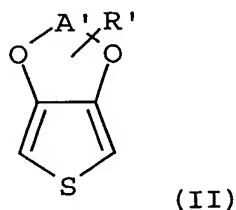
According to a third embodiment of the polymer, according to the present invention, the polymer further comprises at least one 3,4-alkylene-dioxythiophenedioxide compound represented by formula (II):

10



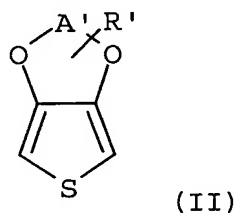
in which: A' represents a C₁₋₅-alkylene bridge; R' represents an optionally substituted C₁₋₂₄-alkyl, C₃₋₁₈-cycloalkyl, C₁₋₁₈-alkoxy or polyethylene oxide group (optionally with at least one substituent
 5 selected from the group consisting of an alcohol, amide, ether, ester or sulphonate group) or an optionally substituted aryl group.

According to a fourth embodiment of the polymer, according to the present invention, the polymer further comprises monomeric units of at least one 3,4-alkylenedioxythiophene compound represented by
 10 formula (II):



in which: A' represents a C₁₋₅-alkylene bridge; R' represents an optionally substituted C₁₋₂₄-alkyl, C₃₋₁₈-cycloalkyl, C₁₋₁₈-alkoxy or polyethylene oxide group (optionally with at least one substituent
 15 selected from the group consisting of an alcohol, amide, ether, ester or sulphonate group), an optionally substituted aryl group or an optionally substituted acyl group.

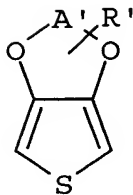
According to a fifth embodiment of the polymer, according to the present invention, the polymer further comprises monomeric units of
 20 at least one 3,4-alkylenedioxythiophene compound represented by formula (II):



in which: A' represents a C₁₋₅-alkylene bridge; and R' represents an optionally substituted C₆₋₂₀-alkyl group.

25 According to a sixth embodiment of the polymer, according to the present invention, the polymer further comprises monomeric units of at least one 3,4-alkylenedioxythiophene compound represented by formula (II):

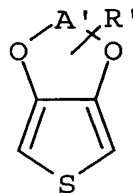
11



(II)

in which: A' represents a C₁₋₅-alkylene bridge; and R' represents an optionally substituted C₈₋₁₈-alkyl group.

According to a seventh embodiment of the polymer, according to the present invention, the polymer further comprises monomeric units of at least one 3,4-alkylenedioxythiophene compound represented by formula (II):



(II)

in which: A' represents a C₁₋₅-alkylene bridge; and R' represents an optionally substituted C₂₋₂₄-alkyl, C₃₋₁₈-cycloalkyl, C₁₋₁₈-alkoxy or an optionally substituted aryl group contains at least one substituent selected from the group consisting of an ether, an ester, an amide, a sulphonate, a phosphonate, a halogen and a hydroxy group.

According to an eighth embodiment of the polymer, according to the present invention, the polymer is a copolymer of 3,4-ethylenedioxythiophene with a 3,4-alkylenedioxythiophenedioxide compound.

According to a ninth embodiment of the polymer, according to the present invention, the polymer is a copolymer of 3,4-ethylenedioxythiophene with 2,3-dihydro-thieno[3,4-b][1,4]dioxine 6,6-dioxide.

According to a tenth embodiment of the polymer, according to the present invention, the polymer is a copolymer of at least one 3,4-alkylenedioxythiophene compound with at least one 3,4-alkylenedioxythiophenedioxide compound and the at least one 3,4-alkylenedioxythiophene compound is selected from the group consisting of: 3,4-ethylenedioxythiophene, 3,4-propylenedioxythiophene, 3,4-dihydro-2H-thieno[3,4-b][1,4]dioxin-2-yl)methanol, 3,4-dihydro-2H-thieno[3,4-b][1,4]dioxepin-3-ol, (2,3-dihydro-thieno[3,4-b][1,4]dioxin-2-yl-methoxy)-acetic acid, 2-{2-[2-(2-methoxy-ethoxy)-ethoxy]-ethoxymethyl}-2,3-dihydro-thieno[3,4-b][1,4]dioxine and 4-(2,3-dihydro-thieno[3,4-b][1,4]dioxin-2-ylmethoxy)-butane-1-sulphonic acid sodium salt.

Thiophene compounds according to formula(II), according to the present invention, can be prepared by known methods such the transesterification reaction disclosed in DE 3804522 and in HOUBEN-WEYL, volume VI/3, part 3, pages 171-173 (1971) using a thiophene derivative such as 3,4-dimethoxythiophene, or the double Williamson reaction as disclosed in 1994 in Electrochimica Acta in volume 39, pages 1345-1347 using a thiophene derivative such as the dimethyl ester of 3,4-dihydroxythiophene-2,5-dicarboxylic acid.

Homopolymers of 3,4-alkylenedioxythiophenedioxide compounds could not be prepared by the inventors by electrochemical or chemical polymerization, but could be prepared from homopolymers of 3,4-alkylenedioxythiophene compounds by oxidation with oxidation with oxidizing agents such as meta-chloroperbenzoic acid, see INVENTION EXAMPLES 6-21.

15

Aqueous dispersions

According to a first embodiment of the aqueous dispersion, according to the present invention, the polymer further comprises monomeric units of at least one thiophene or pyrrole compound.

According to a second embodiment of the aqueous dispersion, according to the present invention, the polymer further comprises monomeric units of at least one 3,4-alkylenedioxythiophene compound.

According to a third embodiment of the aqueous dispersion, according to the present invention, the polyanion is poly(styrenesulphonate).

Polyanion compound

The polyanion compounds for use in the dispersion according to the present invention are disclosed in EP-A 440 957 and include polymeric carboxylic acids, e.g. polyacrylic acids, polymethacrylic acids, or polymaleic acids and polysulphonic acids, e.g. poly(styrenesulphonic acid). These polycarboxylic acids and polysulphonic acids can also be copolymers of vinylcarboxylic acids and vinylsulphonic acids with other polymerizable monomers, e.g. acrylic acid esters, methacrylic acid esters and styrene.

Chemical polymerization process

40

Aspects of the present invention are also provided by a

chemical polymerization process for preparing an aqueous dispersion of a polymer comprising at least one 3,4-alkylenedioxythiophene-dioxide compound and a polyanion comprising the steps of: (i) providing a solution of the polyanion; (ii) adding a 3,4-
5 alkylenedioxythiophenedioxide compound and a thiophene or pyrrole compound to the solution provided in step (i); and (iii) adding an oxidizing or reducing system to the mixture provided in step (ii).

According to a first embodiment of the chemical polymerization process, according to the present invention, the thiophene or
10 pyrrole compound is a 3,4-alkylenedioxy-thiophene compound.

According to a second embodiment of the chemical polymerization process, according to the present invention, inexpensive and easily accessible oxidation agents such as iron(III) salts such as FeCl_3 , the iron(III) salts of organic acids, e.g. $\text{Fe}(\text{OTs})_3$, H_2O_2 , $\text{K}_2\text{Cr}_2\text{O}_7$,
15 alkali and ammonium persulphates, alkali perborates and potassium permanganate are used in the oxidative polymerization.

Theoretically the oxidative polymerization of thiophenes requires 2.25 equivalents of oxidation agent per mole thiophene of formula (I) [see e.g. J. Polymer Science Part A Polymer Chemistry, volume 26, pages 1287-1294 (1988)]. In practice an excess of 0.1 to
20 2 equivalents of oxidation agent is used per polymerizable unit. The use of persulphates and iron(III) salts has the great technical advantage that they do not act corrosively. Furthermore, in the presence of particular additives oxidative polymerization of the
25 thiophene compounds according to formula (I) proceeds so slowly that the thiophenes and oxidation agent can be brought together as a solution or paste and applied to the substrate to be treated. After application of such solutions or pastes heating of the coated substrate accelerates the oxidative polymerization as disclosed in
30 US 6,001,281 and WO 00/14139 herein incorporated by reference.

Reductive polymerization can be performed using the Stille (organotin) or Suzuki (organoboron) routes described in 2002 by Apperloo et al. in Chem. Eur. Journal, volume 8, pages 2384-2396, and as disclosed in 2001 in Tetrahedron Letters, volume 42, pages
35 155-157 and in 1998 in Macromolecules, volume 31, pages 2047-2056 respectively or with nickel complexes as disclosed in 1999 in Bull. Chem. Soc. Japan, volume 72, page 621 and in 1998 in Advanced Materials, volume 10, pages 93-116.

Electroconductive pattern

Aspects of the present invention are provided by an electroconductive pattern comprising a polymer containing 3,4-alkylenedioxythiophenedioxide monomer units.

3,4-alkylenedioxythiophene monomer units can be oxidized to 3,4-alkylenedioxythiophen-1,1-dioxide units by oxidation by peroxides such as m-chloroperoxybenzoic acid (m-CPBA), dimethyldioxirane and $\text{HOF} \cdot \text{CH}_3\text{CN}$ [see references in Journal of Organic Chemistry, volume 63, pages 5497-5506 (1998)].

According to a first embodiment of the electroconductive pattern, according to the present invention, the electroconductive pattern is obtainable using a process comprising the steps of: providing a layer of a polymer containing 3,4-alkylenedioxythiophene monomer units; and pattern-wise oxidizing the polymer containing 3,4-alkylenedioxythiophene monomer units to polymers containing 3,4-alkylenedioxythiophenedioxide monomer units. A suitable oxidizing agent is meta-chloroperbenzoic acid, also known as meta-chloroperoxybenzoic acid.

Support

A transparent or translucent support suitable for use with the electroconductive or antistatic layers, according to the present invention, may be transparent or translucent, rigid or flexible and consist of paper, polymer film, glass, ceramic, a glass-polymer laminate, a polymer laminate, a thermoplastic polymer or a duroplastic polymer. Examples of thin flexible supports are those made of a cellulose ester, cellulose triacetate, polypropylene, polycarbonate and polyester, with poly(ethylene terephthalate) or poly(ethylene naphthalene-1,4-dicarboxylate) being particularly preferred.

Industrial application

Polymers comprising monomeric units of at least one 3,4-alkylenedioxythiophenedioxide compound exhibit high visible light transmittances, good processability and similar work functions but reduced electrical conductivities compared with PEDOT/PSS. Such thiophene polymers can be applied to a wide variety of rigid and flexible substrates and are particularly suitable for flexible substrates such as plastic sheeting, which can be substantially bent

and deformed without the thiophene polymer layer losing its electrical conductivity.

Such thiophene polymers can, for example, be utilized in photovoltaic devices, batteries, capacitors and organic and inorganic electroluminescent devices, in electromagnetic shielding layers, in heat shielding layers, in antistatic coatings for a wide variety of products including photographic film, thermographic recording materials and photothermographic recording materials, in smart windows, in electrochromic devices, in sensors for organic and bio-organic materials, in field effect transistors, in printing plates, in conductive resin adhesives and in free-standing electrically conductive films [see also chapter 10 of Handbook of Oligo- and Polythiophenes, Edited by D. Fichou, Wiley-VCH, Weinheim (1999)].

15

The invention is illustrated hereinafter by way of comparative and invention examples. The percentages and ratios given in these examples are by weight unless otherwise indicated.

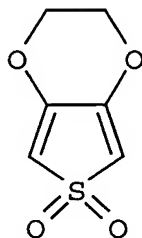
20

SYNTHESIS OF 3,4-ALKYLENEDIOXYTHIOPHENE COMPOUNDS

Synthesis of 2,3-dihydro-thieno[3,4-b][1,4]dioxine 6,6-dioxide (M1)

A solution of 3,4-ethylenedioxythiophene (EDOT, 20.0 g, 0.14 mol) in methylene chloride (140 mL) was cooled to 0°C, blanketed by nitrogen. m-Chloroperoxybenzoic acid (ca. 70%, 104 mL, 3.0 equivalents) was dissolved in methylene chloride (850 mL) and this solution was added dropwise to the EDOT solution over a period of 90 min. After addition the reaction mixture was stirred for another 30 minutes at 25°C after which 2,2'-thiodiethanol (9.5 mL) was added. Stirring was continued for another 10 minutes. Then sodium bicarbonate (40 g) in deionized water (500 mL) was added to the reaction mixture and the two-phase system was vigorously shaken. The organic phase was separated, washed with another portion of aqueous sodium bicarbonate solution, washed with hydrochloric acid (1M) and finally washed with a saturated aqueous sodium chloride solution. The resulting organic solution was then filtered over a layer of silica gel and concentrated. Subsequently methanol was added and the precipitated product was filtered. Finally, the product was recrystallized from ethanol/deionized water (60/40) resulting in pure 2,3-dihydro-thieno[3,4-b][1,4]dioxine 6,6-dioxide as

demonstrated by NMR spectroscopy, mass spectroscopy and FT-IR spectroscopy.



5 PREPARATION OF 3,4-ALKYLENEDIOXYTHIOPHENE-HOMOPOLYMERS

COMPARATIVE EXAMPLE 1

At 25°C, 562.5g of a 5.6% by weight aqueous solution of
10 poly(styrenesulphonic acid) [PSS] (M_n = ca. 40,000; M_w = 290,000;
degree of sulphonation > 90%), 2437.5g of deionized water and 12.78
g (90mmol) EDOT were mixed in a 4L reaction vessel equipped with a
stirrer. 0.225g $Fe_2(SO_4)_3 \cdot 9H_2O$ and 25.7g $Na_2S_2O_8$ were then added to
initiate the polymerization reaction. The reaction mixture was
15 stirred at 25°C for 7h, after which a further 4.3g of $Na_2S_2O_8$ was
added. After an additional reaction time of 16h the reaction
mixture was treated twice with ion exchanger (300mL Lewatit™ S100MB
+ 500mL Lewatit™ M600MB from BAYER). The resulting mixture was
additionally thermally treated at 95°C for 2h and the resulting
20 viscous mixture treated with high shear (microfluidizer at 600Bar).
1800g of a 1.09% by weight blue dispersion of P1 resulted.

COMPARATIVE EXAMPLE 2

25 At 25°C, 438.23g of a 5.99% by weight aqueous solution of
poly(styrenesulphonic acid) [PSS] (M_n = ca. 40,000; M_w = 290,000;
degree of sulphonation > 90%) and 2061.77g deionized water were
mixed in a 4L reaction vessel equipped with a stirrer and a nitrogen
inlet. After bubbling nitrogen through this mixture for 30 minutes,
30 12.78 g (90mmol) of EDOT were added to this solution. 0.225g
 $Fe_2(SO_4)_3 \cdot 9H_2O$ and 25.7g $Na_2S_2O_8$ were then added to initiate the
polymerization reaction. The reaction mixture was stirred at 25°C
for 7h, after which a further 4.3g of $Na_2S_2O_8$ was added. After an
additional reaction time of 16h the reaction mixture was treated
35 twice with ion exchanger (300mL Lewatit™ S100MB + 500mL Lewatit™
M600MB from BAYER). The resulting mixture was additionally

thermally treated at 95°C for 2h and the resulting viscous mixture treated with high shear (microfluidizer at 600Bar). This procedure yielded 1950g of a 1.02% by weight blue dispersion of P2.

COMPARATIVE EXAMPLE 3

At 25°C, 438.23g of a 5.99% by weight aqueous solution of poly(styrene sulphonic acid) [PSS] (M_n = ca. 40,000; M_w = 290,000; degree of sulphonation > 90%) were mixed with 2061.77g of deionized water in a 4L reaction vessel equipped with a stirrer and a nitrogen inlet. After bubbling nitrogen purging through this mixture for 30 minutes, 12.78 g (90mmol) of EDOT was added. 0.225g $Fe_2(SO_4)_3 \cdot 9H_2O$ and 25.7g $Na_2S_2O_8$ were then added to initiate the polymerization reaction. The reaction mixture was stirred at 25°C for 7h, after which a further 4.3g of $Na_2S_2O_8$ was added. After an additional reaction time of 16h the reaction mixture was treated twice with ion exchanger (300mL Lewatit™ S100MB + 500mL Lewatit™ M600MB). The resulting mixture was additionally thermally treated at 95°C for 2h and the resulting viscous mixture was treated with high shear (microfluidizer at 600Bar). This procedure yielded 1840g of a 1.03% by weight blue dispersion of P3.

Preparing electroconductive layers with dispersions based on the dispersions of COMPARATIVE EXAMPLES 1 to 3

25

Coating dispersions were produced by adding 3-glycidoxypyrrol-trimethoxysilane, ZONYL™ FS0100, a copolymer latex of vinylidene chloride, methyl acrylate and itaconic acid (88/10/2) and N-methyl pyrrolidinone to the dispersions of COMPARATIVE EXAMPLES 1 to 3 so as to produce layers, upon doctor blade-coating onto a subbed 175µm poly(ethylene terephthalate) support and drying at 45°C for 3.5 minutes, with the following composition:

PEDOT	28.9mg/m ²
[PEDOT)/PSS	100 mg/m ²]
ZONYL™ FS0100	8 mg/m ²
3-glycidoxypyrrol-trimethoxysilane	100mg/m ²
copolymer latex of vinylidene chloride, methyl acrylate and itaconic acid (88/10/2)	100mg/m ²
N-methyl pyrrolidinone	2mL/m ²

35

Characterization of electroconductive layers prepared with dispersions based on dispersions of COMPARATIVE EXAMPLES 1 to 3

The optical density of the layers was determined by measuring a stack of 10 strips in transmission with a MacBeth™ TD904 densitometer using a visible filter and then obtaining therefrom the optical density of a single strip. The values given in Table 1 include the optical density of the PET-support.

The surface resistance of the layers was measured in a room conditioned to a temperature of 25°C and 30% relative humidity by contacting the printed layer with parallel copper electrodes each 35 mm long and 35 mm apart capable of forming line contacts, the electrodes being separated by a Teflon™ insulator. This enabled direct measurement of the surface resistance to be realized. The results are also given in Table 1.

The layers were then exposed to artificial sunlight (provided by a xenon lamp) through a glass filter in an Atlas Material Testing Technology BV, SUNTEST™ CPS apparatus according to DIN 54 004. The factor given in Table 1 is the ratio of surface resistance after x hours Suntest™ exposure to the surface resistance before the Suntest exposure.

Table 1:

Comparative Example nr	Condition of reaction medium prior to initiator addition	PEDOT/PSS		Initial surface resistance [Ohm/square]	O.D.	Ratio of surface resistance after 48h Suntest™ exposure to initial surface resistance
		type	concentration [wt%]			
1	no oxygen exclusion	P1	1.09	2900	0.067	83
2	purged of oxygen by N ₂ bubbling	P2	1.02	1200	0.066	13
3	purged of oxygen by N ₂ bubbling	P3	1.03	1200	0.065	12

The results in Table 1 show that the initial surface resistance and the stability of the PEDOT/PSS-layers is strongly dependent upon the conditions under which the polymerization is initiated, driving off oxygen by bubbling through with nitrogen resulting in lower surface resistance and higher stability to 48h Suntest™ exposure as shown by lower ratios of surface resistance after Suntest™ exposure to the initial surface resistance.

PREPARATION OF COPOLYMERS OF 3,4-ALKYLENEDIOXYTHIOPHENE AND 3,4-
ALKYLENEDIOXYTHIOPHENEDIOXIDE COMPOUNDS

5

INVENTION EXAMPLES 1 to 5

Dispersions of the copolymers of 3,4-alkylenedioxythiophene and 3,4-alkylenedioxythiophenedioxide compounds of INVENTION EXAMPLES 1 to 5 were prepared by mixing 39.5 g of a 6.01% by weight aqueous solution of poly(styrenesulphonic acid) [PSS] (M_n = ca. 40,000; M_w = 290,000; degree of sulphonation > 90%) with 170.5 g of deionized water and the quantities of EDOT and M1 given in Table 2 at 25°C in a 500 mL reaction vessel equipped with a stirrer. This mixture was homogenized for 30 minutes with an ULTRATURRAX™ at 1000 rpm. 0.015 g $Fe_2(SO_4)_3$ and 1.79 g $Na_2S_2O_8$ were then added to initiate the copolymerization reaction. The reaction mixture was stirred at 25°C for 7h, after which a further 0.3g of $Na_2S_2O_8$ was added. After an additional reaction time of 16h the reaction mixture was treated twice with ion exchanger (21 mL LEWATIT™ S100MB + 35 mL LEWATIT™ M600MB from BAYER). The resulting mixture was additionally thermally treated at 95°C for 2h. This procedure yielded a dispersion of the copolymer (for type, quantity produced and concentrations of copolymer in the dispersion see Table 2).

25 Table 2:

INVENTION EXAMPLE NUMBER	1	2	3	4	5
EDOT wt [g]	0.9	0.5	0.99	0.98	0.95
EDOT [mmoles]	6.33	3.52	6.96	6.89	6.68
Comonomer	M1	M1	M1	M1	M1
Comonomer wt [g]	0.1	0.5	0.01	0.02	0.05
Comonomer [mmoles]	0.48	2.42	0.048	0.097	0.24
Copolymer dispersion	CP1	CP2	CP3	CP4	CP5
wt of (co)polymer dispersion prepared [g]	123	79	135	140	132
(co)polymer concentration in dispersion [wt %]	1.02	1.03	1.49	1.40	1.41

Characterization of the homopolymers and copolymers of INVENTION EXAMPLES 1 to 5 and COMPARATIVE EXAMPLES 1 to 3:

30

The molecular weights of the homopolymers and copolymers of INVENTION EXAMPLES 1 to 5 and COMPARATIVE EXAMPLES 1 to 3 were determined by aqueous gel permeation chromatography relative to

sodium poly(styrene sulphonate) with UV-vis absorption detection at 785 nm.

The particle size of the copolymer/PSS latex particles in the dispersions were determined with a CPS Disc Centrifuge in which particle size distributions are determined using differential centrifugal sedimentation. Particles settle in a fluid under centrifugal field according to Stokes' Law. Sedimentation velocity increases as the square of the particle diameter, so particles that differ in size by only a few percent settle at significantly different rates. In differential sedimentation, all the particles in a sample begin sedimentation as a thin band. A sample of particles is placed on top of a column of clear liquid at the start of the analysis and the particles settle down in the centrifugal field. The detector initially reads maximum intensity, but the signal is reduced when particles reach the detector beam. The reduction in intensity indicates the concentration of particles in the detector beam. When a monochromatic light source is used, Mie theory light scattering can be applied to the intensity data to calculate the particle concentration. When all the particles have passed the detector, the signal returns to the original level. A plot of the particle concentration against the calculated particle diameter provides a differential distribution.

The molecular weights of the copolymers and PEDOT prepared in reaction media purged of oxygen by bubbling through with nitrogen prior to the addition of initiator together with their concentrations in the dispersions produced, the theoretical concentration in mol% in the comonomer and mean particle sizes are given in Table 3.

Table 3:

(Co)polymer nr.	Comonomer		Concentration of copolymer/PSS [wt%]	mean particle size [nm]	molecular weight [785 nm]
	Nr.	mol%			
P1	-	0	1.02	45	550,000
P2	-	0	1.02	47	490,000
P3	-	0	1.03	42	390,000
CP1	M1	7.05	1.02	52	260,000
CP2	M1	40.7	1.03	63	104,000
CP3	M1	0.68	1.49	138	230,000
CP4	M1	1.39	1.40	120	220,000
CP5	M1	3.47	1.41	88	230,000

PREPARATION OF LAYERS PREPARED WITH DISPERSIONS CONTAINING THE COPOLYMERS OF INVENTION EXAMPLES 1 to 5

Coating dispersions were prepared with the dispersions of INVENTION
5 EXAMPLES 1 to 5 as described above for the dispersions of
COMPARATIVE EXAMPLES 1 to 3 so as to produce layer numbers 4 to 8,
upon doctor blade-coating onto a subbed 175µm poly(ethylene
terephthalate) support and drying at 45°C for 3.5 minutes, with the
following composition:

10	Copolymer of EDOT and M1 (or PEDOT)	28.9mg/m ²
	[copolymer of EDOT and M1 (or PEDOT)/PSS	100 mg/m ²]
	ZONYL™ FSO100	8 mg/m ²
	3-glycidoxypopyl-trimethoxysilane	100mg/m ²
	copolymer latex of vinylidene chloride, methyl acrylate	100mg/m ²
	and itaconic acid (88/10/2)	
	N-methyl pyrrolidinone	2mL/m ²

The surface resistance and optical density and the light stability
of layer numbers 4 to 8 were determined as described above for layer
numbers 1 to 3. The results are given in Table 4 together with
15 those for layers 1 to 3.

Table 4:

Layer Nr	(Co) - polymer nr.	Comonomer		Layer containing (co)polymer	
		nr.	mol%	Surface resistance [ohm/square]	O.D.
1	P1			3400	0.06
2	P2	-	0	1200	0.066
3	P3	-	0	1200	0.065
4	CP1	M1	7.05	3×10^5	0.06
5	CP2	M1	40.7	$> 10^{14}$	0.05
6	CP3	M1	0.68	2.6×10^5	-
7	CP4	M1	1.39	1.8×10^5	-
8	CP5	M1	3.47	7.7×10^5	-

20 Preparation of light-emitting devices 1 to 3 with BAYTRON™ P EL and
the copolymers of INVENTION EXAMPLES 1 and 2

Preparation of the Electroluminescent ZnS:Cu dispersion

25 The following solutions were prepared:

Solution 1	zinc acetate dihydrate copper acetate monohydrate deionized water	131,7 g 0,54 g to 600 ml
Solution 2	sodium sulphide nonahydrate ammonia (50%) deionized water	113,6 g 5 ml to 600 ml
Solution 3	TRI* deionized water sodium chloride	40 g to 1000 ml 58,44 g

*TRI = (5-methyl-1,2,4-triazolo-(1,5-a)-pyrimidine-7-ol)

Solutions 1 and 2 at room temperature were added simultaneously both at a flow rate of 500 ml/min to solution 3, held at room temperature and stirred at 1500 rpm. To 1000 ml of the resulting dispersion, 1000 ml of a 1% polyphosphoric acid solution adjusted to pH 6 with ammonium hydroxide was added and the dispersion was concentrated to 1000 ml by means ultrafiltration using a Fresenius F60 cartridge. This dispersion was subsequently diafiltrated by using 5500 ml of a 1% solution of polyphosphoric acid solution adjusted to pH 6 with ammonium hydroxide solution in water. The dispersion was further concentrated to a volume of about 570 ml to produce an aqueous dispersion at pH 6 containing 30g/L ZnS:Cu and 1% polyphosphoric acid. The dispersion was then ball-milled for 1 hour. 20 ml of the resulting dispersion was then ultrasonically treated with a ultrasound bar (Vibra cell VCX 400 W from Sonics & Materials Inc. - amplitude about 78% - output 40%) for 3 minutes while cooling in ice. 3.2 g of a 5% by weight aqueous solution of poly(vinylpyrrolidinone) in water was then added to 16.8 g of the ZnS:Cu dispersion followed by further ultrasonic treatment with the ultrasound bar for 5 minutes. 0.5 ml of ZONYL™ FS0100 was then added as a 1% by weight aqueous solution followed by thorough stirring. The resulting dispersion was then filtered through a 5 µm MILLIPORE™ filter, a nano-dispersion of electroluminescent ZnS:Cu being thereby produced.

Preparation of a patterned ITO electrode:

An indium tin oxide [ITO] layer on 175 µm thick poly(ethylene terephthalate) [PET] from IST, with a surface resistance of about 80 Ohm/square, was used as the hole-conducting electrode. ITO/PET sheets of 5 x 5 cm² were taped off in the middle of the plate with a 2 cm Magic tape Scotch 810 from 3M. The sides of the ITO plates were

then etched with a solution consisting of 50 mL of concentrated hydrochloric acid, 50 mL of deionized water and 4 mL of concentrated nitric acid. After the etching, the ITO/PET sheets were rinsed with water several times and subsequently dried with a hair dryer, dried,
 5 then put into a vessel with isopropanol which was itself put into an ultrasound bath for 10 minutes and finally dried at 50°C for 10 minutes. Each ITO/PET sheet contained a band of 2 cm of conductive ITO in the middle.

10 Preparation and application of the hole injecting and electron blocking layer:

The copolymer dispersions CP06 and CP07, with M1-EDOT-copolymers containing 7.05 and 40.7 mol% M1 respectively, were prepared
 15 analogously to those of CP01 and CP02, but each with concentrations of copolymer/PSS of 1.2% by weight instead of 1.02 and 1.03% by weight respectively.

Dispersion numbers 1 to 3, given in Table 5, were then prepared using BAYTRON™ P EL from BAYER containing 1.2% by weight of
 20 PEDOT/PSS with a PEDOT:PSS weight ratio of 1:2.4 (= P4) and copolymer dispersions CP06 and CP07.

Table 5:

Disper- sion nr	(Co)-polymer dispersion		quantity of 5% aqueous sol- ution of ZONYL® FSO100 [mL]	pH of dis- persion
	nr.	quantity [g]		
1	P4	7.5	3	1.98
2	CP6	7.5	3	2.01
3	CP7	7.5	3	2.10

25 Solutions 1, 2 and 3 were spincoated onto patterned ITO/PET sheets for 6s at 800 rpm followed by 50s at 1500 rpm. The resulting layers were dried at 50°C for 10 minutes giving a layer thickness of about 100 nm thereby producing the substrates 1, 2 and 3 respectively.

30 The thicknesses were measured with a DEKTAK™ profilometer with the layer spincoated on glass. Similar layer thicknesses can be expected on ITO/PET.

Coating with the ZnS:Cu electroluminescent dispersion:

The above-described electroluminescent nano ZnS:Cu-dispersion was spincoated onto substrates 1, 2 and 3 respectively for 6 seconds at 1000 rpm followed by 50 seconds at 2000 rpm followed by drying at 50°C for 10 minutes, thereby producing electroluminescent ZnS:Cu-
5 layers with a thickness of 100 nm.

Application of the aluminium electrode:

Subsequently, a 160 nm thick aluminium electrode (cathode) was
10 vacuum deposited on the spincoated electroluminescent ZnS:Cu-layers at a vacuum of 1.33×10^{-4} N m⁻² Pa using a mask to produce light emitting devices 1 to 3 respectively with the device construction shown in Figure 1. The emission area was 25 mm².

15 Performance of light emitting devices 1 to 3:

The electroluminescence results of light emitting devices 1 to 3 are given in Table 6. With forward bias, the devices exhibited electroluminescence with a λ_{max} of 490 nm. For the lifetime
20 measurements, a forward bias was applied and the voltage was increased so as to maintain the light output constant at ca. 0.5 Cd/m². The maximum applied voltage was 12 V. The lifetime of the light emitting device was taken to be the time between the application of the optimum voltage and the moment no further
25 electroluminescence could be observed. The optimum voltage was that voltage at which maximum light output was observed.

Table 6:

Device nr.	Hole injecting and electron blocking layer	Optimum voltage (V)	Electroluminescence lifetime [s]
1 (Comparison)	P4 + ZONYL™ FSO100	6.9	390
2 (Invention)	CP6 (EDOT-M1-copolymer with a molar ratio of 92.95:7.05) + ZONYL™ FSO100	6.5	675
3 (Invention)	CP7 (EDOT-M1-copolymer with a molar ratio of 59.3:40.7) + ZONYL™ FSO100	5.5	860

30

From Table 6 it can be concluded that the optimum voltage for the block copolymer of PEDOT-PEDOT-dioxide was lower compared with the BAYTRON® P EL hole transporting material, which is normally used.

This probably accounts for the longer electroluminescence lifetimes observed, because a lower mean voltage is necessary to maintain the light output at a constant level.

- 5 Preparation of light-emitting devices 4 to 6 with BAYTRON® P EL and the copolymers of INVENTION EXAMPLES 1 and 2 and an ORGACON™ EL300-anode

In the light-emitting devices 4 to 6, the transparent conductive
10 film ORGACON™ EL300 [a coating comprising PEDOT/PSS on a PET substrate], commercially available from AGFA-GEVAERT N.V., was used as an anode instead of ITO-PET. The ORGACON™ EL300 foil was cut into 5 cm x 5 cm samples. The resist AZ7217 (from CLARIANT) was then spincoated onto the PEDOT/PSS-coating and baked for 30 minutes at
15 90°C giving a dry layer thickness of 2-3 µm. The ORGACON™ EL300 film coated with AZ7217 was then exposed with UV-light of intensity 4mW/cm² through a mask for ca. 60s, in order to illuminate the middle part (5 cm x 2 cm) of the resist to obtain the same pattern as for the ITO-PET described above, followed by a prebake for 45
20 seconds at 110°C and 40s development in the AZ351B developer (from CLARIANT). In order to obtain good adhesion a 30 minute pre-exposure bake at 110°C was necessary. The areas from which the resist was thereby removed were deactivated by dipping the sample in a 1% NaClO solution for 20 seconds. This step was followed by rinsing with
25 water and drying. The foils were then integrally exposed to UV-light with an intensity of 4 mW/cm² for 60s and the resist removed by developing in isopropanol for 90 seconds. The foils were then washed with water and dried. The resulting foils consisted of a conductive area (ca. 300 Ω/square) in the middle of the material with
30 dimensions 5 cm x 2 cm.

The thus patterned ORGACON™ EL300 foils were coated with above-described dispersions 1 to 3 respectively for 6s at 800 rpm followed by 50s at 1500 rpm. The resulting layers were dried at 50°C for 10 minutes. This resulted in a layer thickness of about 100 nm thereby
35 producing substrates 4, 5 and 6 respectively. The thicknesses were measured with a DEKTAK™ profilometer with the layer spincoated on glass. Similar layer thicknesses can be expected on PEDOT/PET.

Substrates 4 to 6 were then coated with the electroluminescent nano ZnS:Cu-dispersion and an aluminium cathode as described above
40 for light-emitting devices 1 to 3, thereby producing light-emitting devices 4 to 6 respectively.

The electroluminescence results of light emitting devices 1 to 3 obtained as described for light-emitting devices 1 to 3, except that the voltage was increased so as to maintain the light output constant at ca. 0.1 Cd/m^2 instead of about 0.5 Cd/m^2 , are given in Table 7.

Table 7:

Light emitting device nr.	Hole injecting and electron blocking layer	Optimum voltage (V)	Electroluminescence lifetime [s]
4 (Comparison)	P4 + ZONYL™ FSO100	6.6	720
5 (Invention)	CP6 (EDOT-M1-copolymer with a molar ratio of 92.95:7.05) + ZONYL™ FSO100	6.3	1260
6 (Invention)	CP7 (EDOT-M1-copolymer with a molar ratio of 59.3:40.7) + ZONYL™ FSO100	5.4	not determined

From Table 7 it can be concluded that the optimum voltage for the block copolymer of PEDOT-PEDOT-dioxide was lower compared with the BAYTRON® P EL hole transporting material, which is normally used. This probably accounts for the longer electroluminescence lifetime observed in the case of light-emitting device 5, because a lower mean voltage was necessary to maintain the light output at a constant level.

COMPARATIVE EXAMPLE 4 and INVENTION EXAMPLES 6-21

Oxidation experiments were carried out by adding different quantities of meta-chloroperbenzoic acid (m-CPBA) to 100 mL of a 1.05 wt% aqueous dispersion of PEDOT/PSS and stirring the resulting mixture for 2h at 25°C. The resulting dispersions were then doctor blade-coated in a standard composition to a standard thickness on a subbed PET support and the surface resistance of the resulting layers determined. The quantities of m-CPBA added, the molar quantity with respect to the PEDOT present and the resulting surface resistance are given in Table 8.

These experiments demonstrate the possibilities of producing an electroconductive pattern by patternwise treatment of PEDOT/PSS-layers with an oxidizing agent such as meta-chloroperbenzoic acid.

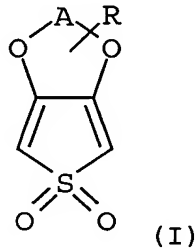
Table 8:

Comparative Example nr	m-CBPA added		Surface resistance at 25°C and 30%RH [Ω /square]
	wt. in [mg]	moles/mole PEDOT	
4	0	0	1.8×10^3
5	5.1	0.01	1.9×10^3
6	10.2	0.02	1.7×10^3
7	25.5	0.05	1.8×10^3
8	51	0.10	2.0×10^3
9	102	0.20	2.6×10^3
10	255	0.50	5.9×10^3
11	510	1.00	2.2×10^4
12	766	1.50	7.8×10^4
13	1021	2.00	4.7×10^5
Invention Example nr.			
6	1530	300	1.0×10^9

The present invention may include any feature or combination of
 s features disclosed herein either implicitly or explicitly or any
 generalisation thereof irrespective of whether it relates to the
 presently claimed invention. In view of the foregoing description
 it will be evident to a person skilled in the art that various
 modifications may be made within the scope of the invention.

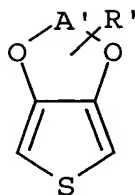
CLAIMS

1. A 3,4-alkylenedioxythiophenedioxide compound.
- 5 2. Compound according to claim 1, wherein said 3,4-alkylenedioxythiophenedioxide compound is represented by formula (I):



- 10 in which: A represents a C₁₋₅-alkylene bridge; R represents an optionally substituted C₁₋₂₄-alkyl, C₃₋₁₈-cycloalkyl, C₁₋₁₈-alkoxy or polyethylene oxide group (optionally with at least one substituent selected from the group consisting of an alcohol, amide, ether, ester or sulphonate group) or an optionally
- 15 substituted aryl group.
3. Compound according to claim 1, wherein said compound is 2,3-dihydro-thieno[3,4-b][1,4]dioxine 6,6-dioxide.
- 20 4. A polymer comprising monomeric units of at least one compound according to claim 1.
5. Polymer according to claim 4, wherein said polymer further comprises monomeric units of at least one thiophene or pyrrole
- 25 compound.
6. Polymer according to claim 4, wherein said polymer further comprises monomeric units of a at least one 3,4-alkylenedioxythiophene compound.
- 30 7. Polymer according to claim 4, wherein said polymer further comprises monomeric units of a at least one 3,4-alkylenedioxythiophene compound represented by formula (II):

29



(II)

in which: A' represents a C₁₋₅-alkylene bridge; R' represents an optionally substituted C₁₋₂₄-alkyl, C₃₋₁₈-cycloalkyl, C₁₋₁₈-alkoxy or polyethylene oxide group (optionally with at least one substituent selected from the group consisting of an alcohol, amide, ether, ester or sulphonate group) or an optionally substituted aryl group.

8. Polymer according to claim 6, wherein said 3,4-alkylenedioxythiophene compound is selected from the group consisting of: 3,4-ethylenedioxythiophene, 3,4-propylenedioxythiophene, 3,4-dihydro-2H-thieno[3,4-b][1,4]dioxin-2-yl)methanol, 3,4-dihydro-2H-thieno[3,4-b][1,4]dioxepin-3-ol, (2,3-dihydro-thieno[3,4-b][1,4]dioxin-2-yl-methoxy)-acetic acid ethyl ester, (2,3-dihydro-thieno[3,4-b][1,4]dioxin-2-yl-methoxy)-acetic acid, 2-{2-[2-(2-methoxy-ethoxy)-ethoxy]-ethoxymethyl}-2,3-dihydro-thieno[3,4-b][1,4]dioxine and 4-(2,3-dihydro-thieno[3,4-b][1,4]dioxin-2-ylmethoxy)-butane-1-sulphonic acid sodium salt.
9. Polymer according to claim 6, wherein said 3,4-alkylene-dioxythiophene compound is 3,4-ethylenedioxythiophene.
10. An aqueous dispersion of a polymer comprising monomeric units of at least one compound according to any of claims 1 to 3 and a polyanion.
11. Aqueous dispersion according to claim 10, wherein said polymer further comprises monomeric units of at least one thiophene or pyrrole compound.
12. Aqueous dispersion according to claim 11, wherein said polymer further comprises monomeric units of at least one 3,4-alkylenedioxythiophene compound.
13. Aqueous dispersion according to claim 10, wherein said polyanion is poly(styrenesulphonic acid).

14. A chemical polymerization process for preparing an aqueous dispersion according to claim 10 comprising the steps of: (i) providing a solution of a polyanion; (ii) adding a compound according to claim 1 and at least one thiophene or pyrrole compound to the solution provided in step (i); and (iii) adding an oxidizing or reducing system to the mixture provided in step (ii).
15. Chemical polymerization process according to claim 14, wherein said thiophene or pyrrole compound is a 3,4-alkylenedioxythiophene compound.
16. Use of an aqueous dispersion according to any of claim 10 for coating an object.
17. A printable paste containing an aqueous dispersion according to claim 10.
18. An electroconductive layer containing a polymer according to claim 4.
19. Use of an electroconductive layer according to claim 18 in a light emitting diode.
20. An antistatic layer containing a polymer according to claim 4.
21. An electroconductive pattern comprising a polymer according to claim 4.
22. A process for preparing said electroconductive pattern according to claim 21 comprising the steps of: providing a layer of a polymer containing 3,4-alkylenedioxythiophene monomer units; and pattern-wise oxidizing said polymer containing 3,4-alkylenedioxythiophene monomer units to polymers containing 3,4-alkylenedioxythiophenedioxide monomer units.

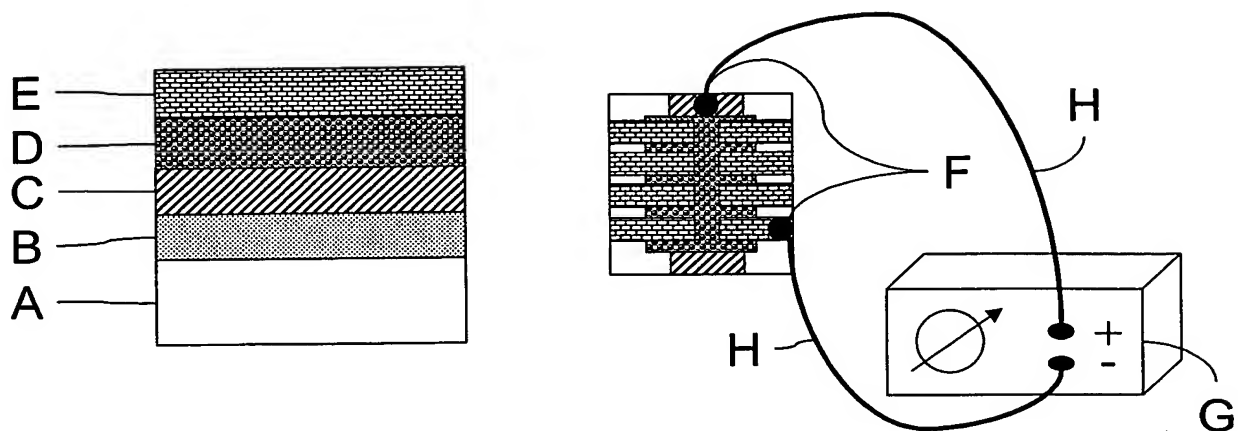
ABSTRACT

NEW 3,4-ALKYLENEDIOXYTHIOPHENEDIOXIDE COMPOUNDS AND POLYMERS
COMPRISING MONOMERIC UNITS THEREOF

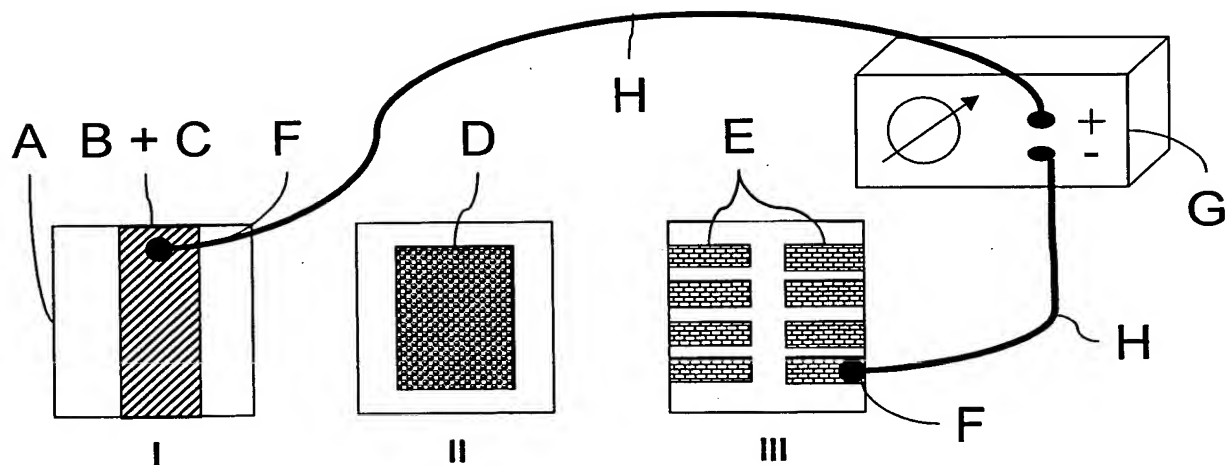
5 A 3,4-alkylenedioxythiophenedioxide compound; a polymer comprising
monomeric units of at least one 3,4-alkylenedioxy-thiophenedioxide
compound; an aqueous dispersion of a polymer comprising monomeric
units corresponding to at least one 3,4-alkylenedioxythiophene-
10 dioxide compound and a polyanion; a chemical polymerization process
for preparing the aqueous dispersion comprising the steps of: (i)
providing a solution of a polyanion; (ii) adding a 3,4-alkylene-
dioxithiophenedioxide compound and a thiophene or pyrrole compound
to the solution provided in step (i); and (iii) adding an oxidizing
15 or reducing system to the mixture provided in step (ii); a printable
paste containing the aqueous dispersion; an electroconductive layer
containing the polymer; the use of the electroconductive layer in a
light emitting diode; an antistatic layer containing the polymer;
and an electroconductive pattern and a process for preparing the
20 electroconductive pattern.

1/1

Figure 1:



5



10

15

20